

Synthesis of $(\eta^5\text{-R}^1\text{C}_5\text{H}_4)(\text{CO})_2\text{Mo}(\eta^2\text{-S}_2\text{R})$ via Reaction of $(\mu\text{-RS})(\mu\text{-}p\text{-MeC}_6\text{H}_4\text{SO}_2\text{S})\text{Fe}_2(\text{CO})_6$ with Nucleophiles $(\eta^5\text{-R}^1\text{C}_5\text{H}_4)(\text{CO})_3\text{MoSeNa}$. Crystal Structure of $(\eta^5\text{-MeO}_2\text{CC}_5\text{H}_4)(\text{CO})_2\text{Mo}(\eta^2\text{-S}_2\text{Bu}^t)$

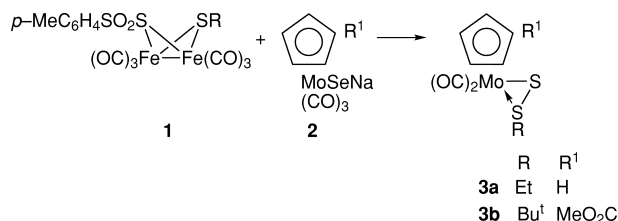
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The reaction of $(\mu\text{-RS})(\mu\text{-}p\text{-MeC}_6\text{H}_4\text{SO}_2\text{S})\text{Fe}_2(\text{CO})_6$ with $(\eta^5\text{-R}^1\text{C}_5\text{H}_4)(\text{CO})_3\text{MoSeNa}$ gave organomolybdenum complexes $(\eta^5\text{-R}^1\text{C}_5\text{H}_4)(\text{CO})_2\text{Mo}(\eta^2\text{-S}_2\text{R})$ (**3a** R = Et, R¹ = H; **3b** R = Bu^t, R¹ = MeO₂C), the structure of **3b** being determined by X-ray diffraction analysis.

We previously reported¹ that the [Et₃NH⁺] salts of sulfur-centered anions $[(\mu\text{-RS})(\mu\text{-S}^-)\text{Fe}_2(\text{CO})_6]$ reacted with toluene-*p*-sulfonyl chloride *p*-MeC₆H₄SO₂Cl to give the expected toluene-*p*-sulfonyl substituted Fe₂S₂ butterfly-shaped clusters of the type $(\mu\text{-RS})(\mu\text{-}p\text{-MeC}_6\text{H}_4\text{SO}_2\text{S})\text{Fe}_2(\text{CO})_6$ (**1**). In order to study the chemical reactivities of this new type of cluster and to further discover applications in the synthesis of transition metal cluster complexes, we recently studied the reaction of clusters **1** with some molybdenum sodium selenides of the type $(\eta^5\text{-R}^1\text{C}_5\text{H}_4)(\text{CO})_3\text{MoSeNa}$ (**2**) prepared by an insertion reaction of the Mo–Na bond in $(\eta^5\text{-R}^1\text{C}_5\text{H}_4)(\text{CO})_3\text{MoNa}$ with selenium powder.² We found that the reaction of an equivalent of clusters **1** (R = Et, Bu^t) with $(\eta^5\text{-R}^1\text{C}_5\text{H}_4)(\text{CO})_3\text{MoSeNa}$ (R¹ = H, MeO₂C) in THF at room temperature afforded two organic monomolybdenum complexes **3a,b**, as shown in Scheme 1.

Complexes **3a,b** are new, and represent, to our knowledge, the first examples of organomolybdenum complexes with an $\eta^2\text{-S}_2\text{R}$ alkylperthio ligand. The IR spectrum of **3a** exhibits two absorption bands at 1940 and 1847 cm⁻¹ for its terminal carbonyls, whereas that of **3b** displays one absorption band at 1713 cm⁻¹ for its ester carbonyl and two absorption bands at 1975 and 1868 cm⁻¹ for its terminal carbonyls. The ¹H NMR spectra of **3a,b** show their respective resonance signals. For example, the parent cyclopentadienyl ring of **3a** exhibits one singlet at δ 6.08, whereas the substituted cyclopentadienyl ring of **3b** displays two sets of signals, one at δ 5.48 assigned to two β -H remote from the substituent



Scheme 1

and the other at δ 5.93–6.15 attributed to two α -H close to the substituent due to the electron-withdrawing effect of the MeO₂C substituent.^{3–5}

In order to confirm the structures of **3a,b**, an X-ray diffraction analysis for **3b** was undertaken. The crystal structure of **3b** was obtained by data collection on a CAD4 diffractometer using Mo-K α radiation ($\lambda = 0.71073$ Å), solved by direct methods and refined by full-matrix least squares. The crystallographic data and data collection

Table 1 Crystal data collection and refinement for **3b**

Formula	C ₁₃ H ₁₆ MoO ₄ S ₂
M _r	396.34
Crystal system	Orthorhombic
Space group	Pbca
a/Å	13.965(3)
b/Å	11.842(2)
c/Å	19.417(2)
V/Å ³	3211(2)
Z	8
D _c /g cm ⁻³	1.640
F(000)	1600
μ /mm ⁻¹	1.085
T/°C	26
Independent reflections	2540
Observed data with $I \geq 3\sigma(I)$	803
Radiation (λ /Å)	Mo-K α (0.710 73)
Scan type	$\omega/2\theta$
$2\theta_{\text{max}}$ /°	46
R	0.064
R _w	0.061
Goodness of fit indicator	0.86
Max. shift in final cycle	0.48
Largest peak in final diff. map/e Å ⁻³	1.96

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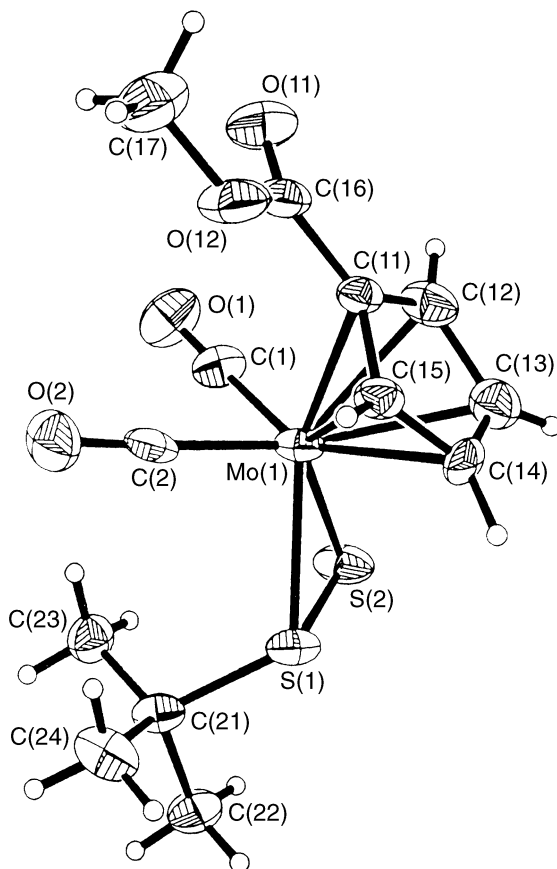


Fig. 1 ORTEP view of **3b** with atom-labeling scheme

parameters are listed in Table 1. An ORTEP view of the molecular structure is shown in Fig. 1.

Fig. 1 shows that the organomolybdenum complex **3b** consists of one molybdenum atom coordinated with one η^5 -MeO₂CC₅H₄, two terminal CO and one η^2 -S₂Bu^t ligands. That is, **3b** contains the same structural unit of (η^5 -MeO₂CC₅H₄)(CO)₂Mo as the complex (η^5 -MeO₂CC₅H₄)(CO)₂Mo[(μ -Ph₂C₂)Co(CO)₃] **4**.⁶ The η^2 -S₂Bu^t in **3b** and [(μ -Ph₂C₂)Co(CO)₃] in **4** are isoelectronic (3e) ligands, which have little influence upon the (η^5 -MeO₂CC₅H₄)(CO)₂Mo structural unit. For instance, the C(11)–C(16) bond length in **3b** and the corresponding one in **4** are 1.456(26) and 1.480(26) Å,⁶ respectively, which lie in between a normal C–C single and normal C=C double bond. This implies that there exists some conjugative interactions between the π -system of the substituent MeO₂C and the cyclopentadienyl ring in both **3b** and **4**. The η^2 -S₂Bu^t ligand of **3b** is coordinated to the Mo(1) atom through its S(1) and S(2) atoms. The Mo(1)–S(1), Mo(1)–S(2) and S(1)–S(2) bond lengths are 2.446(4), 2.506(5) and 2.015(8) Å, respectively. Thus, the three atoms Mo(1), S(1) and S(2) constitute an approximate isosceles triangle. In fact, this coordination geometry resembles those formed from other transition metal and alkylperthio ligands found in some mononuclear transition metal complexes, such as (η^5 -C₅H₅)W(NO)(CH₂SiMe₃)(η^2 -S₂CH₂SiMe₃) [S–S 2.027(4) Å]⁷ and [Os(CO)₂(PPh₃)₂(η^2 -S₂Me)]ClO₄·0.5C₆H₆ [S–S 2.022(7) Å].⁹

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Techniques used: X-Ray diffraction, IR, ¹H NMR

References: 11

Tables 2 and 3: Atomic coordinates and thermal parameters, selected bond lengths and bond angles

Appendix: Tables of anisotropic thermal parameters and complete bond lengths and bond angles

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